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isotope, one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behaviour but ...

In this concept tutorial, learn about what an isotope is, some common isotopes and their uses, and how isotopes form and breakdown.

Isotopes share the same chemical properties but may have different nuclear stability and physical properties. Some isotopes are stable, while others are radioactive and ...

Isotopes are members of a family of an element that all have the same number of protons but different numbers of neutrons. The number of protons in a nucleus determines the element's ...

Isotopes are atoms of the same element that have the same number of protons (i.e., atomic number, "Z") but a different number of neutrons, meaning that their mass number, "A", varies.

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Like everything we see in the world, isotopes are a type of atom, the smallest unit of matter that retains all the chemical properties of an element. Isotopes are forms of a ...

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Explore isotopes, their definition, meaning, characteristics, examples, and various uses in science and industry for a deeper understanding of this concept.

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